

Naming Acids and Bases Worksheet  
Chemistry  
2 points

Name: \_\_\_\_\_

Date: \_\_\_\_\_ Hour: \_\_\_\_\_

- 1) What is a binary acid? \_\_\_\_\_  
\_\_\_\_\_
- 2) What is a monatomic ion? \_\_\_\_\_
- 3) What is a polyatomic ion? \_\_\_\_\_
- 4) Where are the halogens located on the periodic table? \_\_\_\_\_
- 5) Name the following acids:

| Acid formula                   | Positive ion   | Negative ion    | Acid name         |
|--------------------------------|----------------|-----------------|-------------------|
| ex> HCl                        | H <sup>+</sup> | Cl <sup>-</sup> | hydrochloric acid |
| HBr                            |                |                 |                   |
| HNO <sub>3</sub>               |                |                 |                   |
| H <sub>2</sub> SO <sub>4</sub> |                |                 |                   |
| H <sub>3</sub> PO <sub>4</sub> |                |                 |                   |
| H <sub>2</sub> Se              |                |                 |                   |

- 6) Write the formula of the following acids:

| Acid name         | Positive ion   | Negative ion                  | Acid formula                   |
|-------------------|----------------|-------------------------------|--------------------------------|
| ex> carbonic acid | H <sup>+</sup> | CO <sub>3</sub> <sup>2-</sup> | H <sub>2</sub> CO <sub>3</sub> |
| hydrofluoric acid |                |                               |                                |
| citric acid       |                |                               |                                |
| acetic acid       |                |                               |                                |
| hydroiodic acid   |                |                               |                                |

# Introduction To Acids And Bases A Webquest Answers

**Elizabeth R. Kellerman**



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