

COMPUTER APPLICATIONS

(Theory)

(Two hours)

Answers to this Paper must be written on the paper provided separately.

You will not be allowed to write during the first 15 minutes.

This time is to be spent in reading the question Paper.

The time given at the head of this Paper is the time allowed for writing the answers.

This paper is divided into two sections

Attempt all questions from Section A and any four questions from Section B.

The intended marks for questions or parts of questions are given in brackets [].

SECTION - A (40 MARKS)

(Attempt ALL Questions)

Question 1.

- (a) Why is a class called a *factory of objects* ? [2]
- (b) State the difference between a boolean literal and a character literal [2]
- (c) What is the *use and syntax* of a ternary operator ? [2]
- (d) Write one word answer for the following :
 - (i) A method that converts a string to a primitive integer data type.
 - (ii) The *default initial value* of a boolean variable data type. [2]
- (e) State one similarity and one difference between *while* and *for* loop. [2]

Question 2.

- (a) Write the function prototype for the function "sum" that takes an integer variable (x) as its argument and returns a value of float data type. [2]
- (b) What is the use of the keyword *this* ? [2]
- (c) Why is a class known as composite data type ? [2]
- (d) Name the keyword that :
 - (i) is used for allocating memory to an array.
 - (ii) causes the control to transfer back to the method call. [2]
- (e) Differentiate between pure and impure function. [2]

Question 3.

- (a) Write an expression for $\frac{(a+b)^n}{\sqrt{3+b}}$ [2]
- (b) The following is a segment of a program.
`x = 1; y = 1;`
`if (n > 0)`

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Arturo Cuomo



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web cycloalkanes are cyclic hydrocarbons meaning that the carbons of the molecule are arranged in the form of a ring cycloalkanes are also saturated meaning that all of the carbons atoms that make up the ring are single bonded

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web carbon 1 sp 180 carbon 2 sp 180 carbon 3 sp^2 120 carbon 4 sp^2 120 carbon 5 sp^3 109.5 chemically the alkynes are similar to the alkenes since the $C \equiv C$ functional group has two π bonds alkynes typically react even more readily and react with twice as much reagent in addition reactions