

Sun of knowledge school

Grade 8 final exam

chemistry

Date: / /2024

Student Name :

Q1- this equation represent an exothermic reaction



Which change would move the equilibrium to the left ?

(5/___)

A- an increase in the concentration of substance A

B- an increase in the pressure

C-an increase in the temperature

D-removal of substance C from the reaction vessel

Q2- ethanol C₂H₅OH is manufactured from ethane C₂H₄ and steam H₂O

(5/___)



Predict the effect of a catalyst on the position of equilibrium ?

Q3- N₂ + 3H₂ -----> 2NH₃ IT IS AN EXOTHERMIC reaction

Catalyst are not consumed by a reaction . why is the iron catalyst in the Haber process changed regularly ?

(5/___)

Grade8 Final Exam Scope 2014 Caps

Puja Mehta



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