

Name: _____ Date: _____ Period: _____

Atomic Structure, Valence Electrons, and the Periodic Table

Background Info:

Atoms are the smallest unit consist of three main subatomic particles: positively charged particles located with neutral neutrons. Negatively charged electrons orbit the nucleus in shells. Atoms form chemical bonds. Some called valence electrons. This type determining an atom's reactivity.

Periodic Table of Elements:

Label the Periods, Families, and Valence Electrons.

1 H																															
2 He	3 Li																														
4 Be	5 B	6 C	7 N	8 O	9 F	10 Ne																									
11 Na	12 Mg	13 Al	14 Si	15 P	16 S	17 Cl	18 Ar																								
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr														
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe														
55 Cs	56 Ba	57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og

Valence Electrons:

Write the Number of Valence Electrons.

S = _____ F = _____
 H = _____ Cl = _____
 Al = _____ I = _____

Reading Elements on the Periodic Table:

Fill in the Information Below

4
Be
Beryllium
9.0122

← _____, Number of Protons (and Electrons)
 ← _____, Element Abbreviation
 ← _____, Formal Element Name
 ← _____, Equal to Protons + Neutrons

How to Find Protons? → Look at the Atomic Number: Atomic # _____ Protons # _____
 How to Find Electrons? → Look at the Atomic Number: Atomic # _____ Electrons # _____
 How to Find Neutrons? → Take Mass # - Atomic #: Mass # _____ Neutrons # _____

Reading Elements on the Periodic Table:

Fill in the Information Below using the Rounded Mass Number

18
Ar
Argon
39.948

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

15
P
Phosphorus
30.974

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

31
Ga
Gallium
69.723

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

53
I
Iodine
126.90

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

52
Te
Tellurium
127.60

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

55
Cs
Cesium
132.91

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

1
H
Hydrogen
1.008

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

12
Mg
Magnesium
24.305

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

Electrons And Periodicity Packet Answers Unit 3

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