

The Mole Study Guide

Mass Conversions: Round your answers to 2 places after the decimal! Make sure to show your steps!

1) There are 3.75 moles of CO_2 .

a) mass

b) mass

c) mass

2) How many g

a) 24.6

b) 3.04

c) 64.2

3) There are 4

a) mass

b) mass

4) There are 3

a) mass

b) mass

5) There are 3

a) mass

b) mass

Name _____ Period _____ Date _____

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Fill in the blank boxes for the image below.



Calculate the Molar Mass for the following compounds.

1) H_2O

2) H_2S

3) NaCl

4) NaCl

Calculate the Percent Composition for each element in the following compound.

1) H_2O

2) H_2S

3) NaCl

4) NaCl

5) NaCl

6) NaCl

7) NaCl

8) NaCl

Determine the Empirical and Molecular Formula for the compound given the information below.

1) A substance has the following percent composition: 58.80% Hg, 21.80% Cl, 1.40% H. The molar mass for substance is 129.6 g/mol.

a) If you have 100 g of substance, how many grams of Hg, Cl, and H do you have?

b) Determine the Empirical Formula for substance.

c) Determine the Molecular Formula for substance.

Mole Study Guide

Jicheng Xie



Mole Study Guide:

Unveiling the Magic of Words: A Overview of "**Mole Study Guide**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Mole Study Guide**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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