

Name _____ answer key _____ Date _____

Stoichiometry: Mass – Mole Worksheet

Based on the following chemical equation, please answer the following questions!



Calculate the amount of moles of NaCl needed to produce 75 g of FeCl₃.

9 FeCl ₃	1 mole FeCl ₃	mole NaCl
9 FeCl ₃	9 FeCl ₃	mole FeCl

Calculate the amount of Fe₂ (SO₄)₃ in grams produce 215 g Na₂SO₄.

9 Na ₂ SO ₄	1 mole Na ₂ SO ₄	mole Fe ₂ (SO ₄) ₃	9 Fe ₂ (SO ₄) ₃
9 Na ₂ SO ₄	9 Na ₂ SO ₄	mole Na ₂ SO ₄	1 mole Fe ₂ (SO ₄) ₃

Calculate the amount in grams of FeCl₃ produced with 95g of Fe (SO₄)₃.

9 Fe ₂ (SO ₄) ₃	1 mole Fe ₂ (SO ₄) ₃	mole FeCl ₃	9 FeCl ₃
9 Fe ₂ (SO ₄) ₃	9 Fe ₂ (SO ₄) ₃	mole Fe(SO ₄) ₃	1 mole FeCl ₃

Calculate the amount of moles of Fe₂(SO₄)₃ needed to react with 123.5 g of NaCl

9 NaCl	1 mole NaCl	mole Fe(SO ₄) ₃
9 NaCl	9 NaCl	mole NaCl

Based on the following chemical equation, please answer the following questions!



9 Ca(HCO ₃) ₂	1 mole Ca(HCO ₃) ₂	mole CaCO ₃
9 Ca(HCO ₃) ₂	9 Ca(HCO ₃) ₂	mole Ca(HCO ₃) ₂

Mole Short Answer Questions

Julia Schneider



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Table of Contents Mole Short Answer Questions

1. Understanding the eBook Mole Short Answer Questions
 - The Rise of Digital Reading Mole Short Answer Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Mole Short Answer Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mole Short Answer Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mole Short Answer Questions
 - Personalized Recommendations
 - Mole Short Answer Questions User Reviews and Ratings
 - Mole Short Answer Questions and Bestseller Lists
5. Accessing Mole Short Answer Questions Free and Paid eBooks
 - Mole Short Answer Questions Public Domain eBooks
 - Mole Short Answer Questions eBook Subscription Services
 - Mole Short Answer Questions Budget-Friendly Options

6. Navigating Mole Short Answer Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Mole Short Answer Questions Compatibility with Devices
 - Mole Short Answer Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mole Short Answer Questions
 - Highlighting and Note-Taking Mole Short Answer Questions
 - Interactive Elements Mole Short Answer Questions
8. Staying Engaged with Mole Short Answer Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mole Short Answer Questions
9. Balancing eBooks and Physical Books Mole Short Answer Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mole Short Answer Questions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Mole Short Answer Questions
 - Setting Reading Goals Mole Short Answer Questions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mole Short Answer Questions
 - Fact-Checking eBook Content of Mole Short Answer Questions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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