# Mole/Mole Ratio Problems

1. \_\_N₂ + \_\_H₂ → \_\_NH₃

- a. How many moles of hydrogen are needed to completely react with two moles of nitrogen?
- b. How many moles of nitrogen trihydride can be produced with 5 moles of nitrogen?
- c. How many moles of nitrogen are needed to produce .5 moles of nitrogen trihydride?
- d. How many liters of nitrogen trihydride can be produced from 24 liters of nitrogen gas?
  - \_\_KCIO<sub>3</sub> → \_\_KCI + \_\_O<sub>2</sub>
  - a. How many moles of oxygen are produced by the decomposition of six moles of potassium chlorate?
  - b. How many moles of potassium chloride are produced by the decomposition of .75 moles of potassium chlorate?
  - c. How many moles of potassium chlorate are needed to produce 100 moles of oxygen?
  - d. How many liters of oxygen will the decomposition of 100 liters of potassium chlorate produce?

# **Mole Ratios Worksheet Chapter 12**

Stanley I. Sandler

#### **Mole Ratios Worksheet Chapter 12:**

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Holt Chemistry Ralph Thomas Myers, 2004

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