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This question is not based on a passage.

#### Question 4

Which of these statements concerning peptide bonds is FALSE?

- ☐ (A) Their formation involves a reaction between an amino group and a carboxyl group.
- ☐ (B) They are the primary bonds that hold amino acids together.
- ☐ (C) They have partial double bond character.
- ☒ (D) Their formation involves hydrolytic reactions.

⚑ Answer

D

Peptide bonds are the primary covalent bond between the amino acids that make up proteins, making choice (D) incorrect. They involve a condensation reaction between the amino group of one amino acid and the carboxyl group of an adjacent amino acid, eliminating choice (A). The peptide bond has a partial double bond character because the double bond can resonate between C=O and C=N. Thus, the peptide bond has a partial double bond character and exhibits limited rotation, eliminating choice (C). By process of elimination, choice (D) is false. Formation of the peptide bond is a condensation reaction—specifically a dehydration reaction involving the loss of water—not a hydrolytic reaction involving the addition of water.

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# Kaplan Explanations Aamc Test 10

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










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