Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question, using the letters in each "box" to identify the required element(s).

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List elements in alphabetical order when more than 1 element answers a question.

Element A is in a box all by itself because: A) it is really a Noble Gas
B) it is a gas
C) it is NOT a
gas
D) it does NOT share the properties of the other elements in its group
E) it is a liquid

2) Which of these elements is/are highly reactive? A) "c" B) "E" C) "R" D) A, B, and C E) A and B only

3) What is the letter of the element whose atomic number is closest to the element labeled "z"?

4) Which letter labels the halogen element with the lowest atomic number?

5) Which of these elements is the least reactive? A) "o" B) "x" C) "A" D) "o" E) "s"

6) Which of these elements is/are <u>NOT</u> transition elements/metals (list ALL that are not transition elements)? A) "G" B) "N" C) "N" D) "N" E) "Y"

continued on other side

Elements And The Periodic Table Key Terms Answer Key

Raffaela Di Napoli

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Table of Contents Elements And The Periodic Table Key Terms Answer Key

- 1. Understanding the eBook Elements And The Periodic Table Key Terms Answer Key
 - The Rise of Digital Reading Elements And The Periodic Table Key Terms Answer Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Elements And The Periodic Table Key Terms Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - \circ Features to Look for in an Elements And The Periodic Table Key Terms Answer Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Elements And The Periodic Table Key Terms Answer Key
 - Personalized Recommendations
 - Elements And The Periodic Table Key Terms Answer Key User Reviews and Ratings

- Elements And The Periodic Table Key Terms Answer Key and Bestseller Lists
- 5. Accessing Elements And The Periodic Table Key Terms Answer Key Free and Paid eBooks
 - Elements And The Periodic Table Key Terms Answer Key Public Domain eBooks
 - Elements And The Periodic Table Key Terms Answer Key eBook Subscription Services
 - Elements And The Periodic Table Key Terms Answer Key Budget-Friendly Options
- 6. Navigating Elements And The Periodic Table Key Terms Answer Key eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Elements And The Periodic Table Key Terms Answer Key Compatibility with Devices
 - Elements And The Periodic Table Key Terms Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Elements And The Periodic Table Key Terms Answer Key
 - Highlighting and Note-Taking Elements And The Periodic Table Key Terms Answer Key
 - Interactive Elements Elements And The Periodic Table Key Terms Answer Key
- 8. Staying Engaged with Elements And The Periodic Table Key Terms Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Elements And The Periodic Table Key Terms Answer Key
- 9. Balancing eBooks and Physical Books Elements And The Periodic Table Key Terms Answer Key
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Elements And The Periodic Table Key Terms Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Elements And The Periodic Table Key Terms Answer Key
 - Setting Reading Goals Elements And The Periodic Table Key Terms Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Elements And The Periodic Table Key Terms Answer Key
 - Fact-Checking eBook Content of Elements And The Periodic Table Key Terms Answer Key
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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