

# ANSWER KEY

## ATOMS AND ELECTRONS STUDY GUIDE

**Learning Goals:** (a) (a.i) P. 4.1 - Explain a new scientific theory of atomic scale phenomena using theory by describing changes in the atomic model over time including those changes such as the introduction of quantum mechanics. (a.i.ii) P. 4.2 - Explain the scientific theory of atomic scale phenomena using theory by describing the scientific changes in terms of quantum mechanical calculations, including wavefunctions that are not 100% accurate in terms of their mass, electric charges, and locations within the atom. (a.i.iii) P. 4.3 - Explain properties of atoms and their quantum nature including wave functions, wavefunctions of their electrons, orbitals, P. 4.4 - Explain the importance of wavefunctions in the first fundamental principle of quantum mechanics, wave functions, wave functions.

### ANSWER KEY

**Questions:** What are the atomic theory concepts in the current scientific theory?

QUESTION	ANSWER KEY	ANSWER KEY
1. (a)	Classical (a.i.ii)	1. Everything is made of atoms that are made of matter, electric, or magnetic.
2. (a)	Classical (a.i.ii)	2. The atomic theory of matter is mostly a quantum theory of matter, which is a quantum theory of matter.
3. (a)	Classical (a.i.ii)	3. Quantum mechanics (quantum)
4. (a)	Classical (a.i.ii)	4. Group numbers (atomic numbers)
5. (a)	Classical (a.i.ii)	5. Atomic mass
6. (a)	Classical (a.i.ii)	6. Atomic mass (atomic mass)

**Questions:** What are the atomic theory concepts in the current scientific theory?

- 1. (a)** - P. 4.1.ii - The number of protons equals the number of electrons.
- 2. (a)** - P. 4.1.ii - The number of protons equals the number of electrons.
- 3. (a)** - P. 4.1.ii - The number of protons equals the number of electrons.

# Electrons In Atoms Guided Practice Problems Answers

**Arshad Iqbal**



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