

Unit 6 – Worksheet 3

Ionic Compounds

Properties

Basic structural unit

1. Give the name of the following simple binary ionic compounds.

- Na_2O
- K_2S
- MgCl_2
- CaBr_2
- BaI_2
- Al_2S_3
- CaBr
- AgF

2. Give the name of the following simple binary ionic compounds.

- Na_3N
- K_2O
- AgBr
- MgI_2
- SrO

3. Write the formula for the following binary ionic compounds.

- lithium bromide
- sodium iodide
- silver sulfide
- cesium oxide
- beryllium iodide
- barium hydride
- aluminum fluoride
- potassium oxide

4. Write the formula for these ionic substances.

- silver oxide
- aluminum sulfide
- sodium nitride
- barium chloride
- strontium hydride

Modeling Chemistry U6 Ws 5 V1 0

Yan Bai



Modeling Chemistry U6 Ws 5 V1 0:

This book delves into Modeling Chemistry U6 Ws 5 V1 0. Modeling Chemistry U6 Ws 5 V1 0 is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Modeling Chemistry U6 Ws 5 V1 0, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Modeling Chemistry U6 Ws 5 V1 0
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 - Chapter 5: Conclusion
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