

7 One way of establishing a reactivity series is by displacement reactions.

- (a) A series of experiments was carried out using the metals lead, magnesium, zinc and silver. Each metal was added in turn to aqueous solutions of the metal nitrates.

The order of reactivity was found to be:

magnesium	most reactive
zinc	
lead	↓
silver	least reactive

- (b) Complete the table.

✓ = reacts

✗ = does not react

aqueous solution	metal			
	lead Pb	magnesium Mg	zinc Zn	silver Ag
lead(II) nitrate		✓	✓	✗
magnesium nitrate				
zinc nitrate				
silver nitrate				

[3]

- (c) Displacement reactions are redox reactions.

On the following equation, draw a ring around the reducing agent and an arrow to show the change which is oxidation.



[2]

- (d) Complete the following ionic equation.



[1]

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