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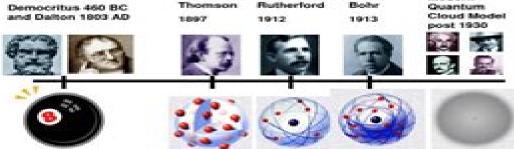
History of the Atom Timeline

Modern.

Dalton's

Atomic Theory





John Dalton (1766 - 1844):

John Dalton was an English chemist. His ideas form the atomic theory of matter. Here are his ideas.

- All elements are composed (made up) of atoms. It is impossible to divide or destroy an atom.
- All atoms of the same elements are alike. (One atom of oxygen is like another atom of oxygen.)
- Atoms of different elements are different. (An atom of oxygen is different from an atom of hydrogen.)
- Atoms of different elements combine to form a compound. These atoms have to be in definite whole number ratios.
 For example, water is a compound made up of 2 atoms of hydrogen and 1 atom of oxygen (a ratio of 2:1). Three atoms of hydrogen and 2 atoms of oxygen cannot combine to make water.
- 1. What is the name of John Dalton's theory? THE ATOMIC THEORY OF MATTER
- 2. What are elements made of? ELEMENTS ARE MADE OF ATOMS
- 3. An atom of hydrogen and an atom of carbon are DIFFERENT
- 4. What are compounds made of? COMPOUNDS ARE MADE OF ATOMS OF TWO OR MORE DIFFERENT ELEMENTS:
- 5. The ratio of atoms in HCl is: a) 1:3 b) 2:1 c) 1:1
- 6. The ratio of atoms in H₂O is: a) 1:3 b) 2:1 c) 1:1 2 atoms of hydrogen and 1 atom of oxygen

History Of The Atom Answersheet

RJ Shavelson

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Embracing the Beat of Phrase: An Mental Symphony within **History Of The Atom Answersheet**

In some sort of eaten by displays and the ceaseless chatter of fast conversation, the melodic splendor and psychological symphony developed by the written term often fade into the background, eclipsed by the persistent sound and disturbances that permeate our lives. But, nestled within the pages of **History Of The Atom Answersheet** a charming literary prize brimming with organic thoughts, lies an immersive symphony waiting to be embraced. Crafted by a wonderful musician of language, that fascinating masterpiece conducts viewers on a psychological trip, well unraveling the concealed melodies and profound affect resonating within each carefully constructed phrase. Within the depths of the moving analysis, we will investigate the book is central harmonies, analyze its enthralling writing model, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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