

Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question. using the letters in each "box" to identify the required element(s).

The diagram shows a crossword puzzle grid with the following letters placed in specific cells:

- Across:** A (1,1), B (1,10), C (2,1), D (2,9), E (2,10), F (3,2), G (3,7), H (3,9), I (3,10), J (4,1), K (4,10), L (4,2), M (4,7), N (4,9), O (4,10), P (4,11), Q (5,1), R (5,2), S (5,7), T (5,8), U (5,9), V (6,1), W (6,9), X (6,1), Y (6,9), Z (6,11).
- Down:** A (1,1), B (1,10), C (2,1), D (2,9), E (2,10), F (3,2), G (3,7), H (3,9), I (3,10), J (4,1), K (4,10), L (4,2), M (4,7), N (4,9), O (4,10), P (4,11), Q (5,1), R (5,2), S (5,7), T (5,8), U (5,9), V (6,1), W (6,9), X (6,1), Y (6,9), Z (6,11).

Arrows point from the letters R and Z to a separate section below the grid, which contains the letters V, W, X, and Y.

List elements in *alphabetical order* when more than 1 element answers a question.

- _____ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid
- _____ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "E" **C)** "K" **D)** A, B, and C **E)** A and B only
- _____ 3) What is the letter of the element whose atomic number is closest to the element labeled "Z"?
- _____ 4) Which letter labels the halogen element with the lowest atomic number?
- _____ 5) Which of these elements is the least reactive? **A)** "J" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"
- _____ 6) Which of these elements is/are NOT transition elements/metals. (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "K" **D)** "N" **E)** "Y"

continued on other side

Groups Of The Periodic Table Answer Key

R Barnett



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Table of Contents Groups Of The Periodic Table Answer Key

1. Understanding the eBook Groups Of The Periodic Table Answer Key
 - The Rise of Digital Reading Groups Of The Periodic Table Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Groups Of The Periodic Table Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Groups Of The Periodic Table Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Groups Of The Periodic Table Answer Key
 - Personalized Recommendations
 - Groups Of The Periodic Table Answer Key User Reviews and Ratings
 - Groups Of The Periodic Table Answer Key and Bestseller Lists
5. Accessing Groups Of The Periodic Table Answer Key Free and Paid eBooks
 - Groups Of The Periodic Table Answer Key Public Domain eBooks
 - Groups Of The Periodic Table Answer Key eBook Subscription Services
 - Groups Of The Periodic Table Answer Key Budget-Friendly Options
6. Navigating Groups Of The Periodic Table Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Groups Of The Periodic Table Answer Key Compatibility with Devices
 - Groups Of The Periodic Table Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Groups Of The Periodic Table Answer Key
 - Highlighting and Note-Taking Groups Of The Periodic Table Answer Key
 - Interactive Elements Groups Of The Periodic Table Answer Key
8. Staying Engaged with Groups Of The Periodic Table Answer Key

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Groups Of The Periodic Table Answer Key
9. Balancing eBooks and Physical Books Groups Of The Periodic Table Answer Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Groups Of The Periodic Table Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Groups Of The Periodic Table Answer Key
- Setting Reading Goals Groups Of The Periodic Table Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Groups Of The Periodic Table Answer Key
- Fact-Checking eBook Content of Groups Of The Periodic Table Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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