

Name Tara ShettyDate 9/24/15 Pd 5**Chemistry – Unit 4 Worksheet 3**

Use the following information about the masses of elements in each pair of compounds to help you suggest formulas that account for these ratios.

1. Compounds of carbon and oxygen

Compound A: 57.1 g O / 42.9 g C

Compound B: 72.7 g O and 27.3 C

$$A = \frac{57.1}{42.9} = 1\frac{1}{3}$$

$$B = \frac{72.7}{27.3} = 2\frac{2}{3}$$

a. Determine the value of the ratio $\frac{\text{mass O}}{\text{mass C}}$ in each compound. A $1\frac{1}{3}$ B $2\frac{2}{3}$

b. How does the mass ratio for compound B compare to that in compound A?

compound B is larger than compound A

c. Express these ratios as improper fractions. $\frac{57.1}{42.9} + \frac{72.7}{27.3} \left(\frac{4}{3} + \frac{8}{3} \right)$

d. For each hypothesis, sketch particle diagrams for the compounds of A and B that account for these mass ratios. Write the formula for the compound in each diagram.

Hypothesis 1

Atoms of C and O have the same mass

A $\frac{4}{3}$ B $\frac{8}{3}$ **Hypothesis 2**

Atoms of O are heavier than C atoms by the ratio in compound A.

A



B



Modeling Chemistry Unit 10 2 Answer

José Teixeira

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Hadlock, 2020-05-05 This manual is meant to provide supplementary material and solutions to the exercises used in Charles Hadlock's textbook *Mathematical Modeling in the Environment*. The manual is invaluable to users of the textbook as it contains complete solutions and often further discussion of essentially every exercise the author presents in his book. This includes both the mathematical/computational exercises as well as the research questions and investigations. Since the exercises in the textbook are very rich in content rather than simple mechanical problems and cover a wide range, most readers will not have the time to work out every one on their own. Readers can thus still benefit greatly from perusing solutions to problems they have at least thought about briefly. Students using this manual still need to work out solutions to research questions using their own sources and adapting them to their own geographic locations or to numerical problems using their own computational schemes, so this manual will be a useful guide to students in many course contexts. Enrichment material is included on the topics of some of the exercises. Advice for teachers who lack previous environmental experience but who want to teach this material is also provided and makes it practical for such persons to offer a course based on these volumes. This book is the essential companion to *Mathematical Modeling in the Environment*. **18 Years Chapterwise**

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