May 2010

SEC Chemistry Paper 1

Marking Scheme

Section A

(a)		ectron -1					
(b			Electrons	Neutrons	Protons	Mass number	
	Λ	.tom R	17	18			
	Lie	on Q 2-			8	16	
	Te	on T *	2			7	
	A	tom W	17			37	56.2
	(iii)	Ranc	i W			both correct if one correct	
(a)		Moles Moles Ratio	carbon = 85. hydrogen = 1 1C: 2H E.F.		Award 0 i		6 mar
(a)		Moles Ratio If E.F	carbon = 85. hydrogen = 1 1C: 2H E.F.	4.29 / 1 = 14.2	Award 0 i	f one correct	6 mar
(a)	(ii)	Moles Moles Ratio If E.F. M.F. =	s carbon = 85. hydrogen = 1 IC : 2H E.F. is given as (C = 3 x EF = C ₃ H ₆	(4.29 / 1 = 14.2 = CH ₂ TH ₂) ₀ deduct ½ CHBrCH ₂ Br	Award 0 i	cept C ₃ H ₆ Br ₂	6 mar
	(ii)	Moles Moles Ratio If E.F. M.F. = E.F. = C ₃ H ₆	s carbon = 85. s hydrogen = 1 IC : 2H E.F. is given as (C = 3 x EF C ₃ H ₆ + Br ₂ © CH ₃ 6	(4.29 / 1 = 14.2 = CH ₂ TH ₂) ₀ deduct ½ CHBrCH ₂ Br	Award 0 i	cept C ₃ H _a Br ₂ ulae award 2 award 1 s orange-brown	6 mar
	(ii) (iii)	Moles Moles Ratio If E.F. M.F. = C ₃ H ₆ Red-o	s carbon = 85. s hydrogen = 1 1C : 2H E.F. is given as (C = 3 x EF = C ₃ H ₆ + Br ₂ © CH ₃ G	(4.29 / 1 = 14.2 = CH ₂ "H ₂) ₀ deduct ½ CHBrCH ₂ Br All If no	Award 0 i	cept C ₃ H ₆ Br ₂ alae award 2 award 1 s orange-brown	

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