#### 8.4 Life on Earth

#### 1. Analysis of the oldest sedimentary rocks provides evidence for the origin of life

Identify the relationship between the conditions on early Earth and the origin of organic molecules (1.2.1)

- Earth is about 4.6 billion years old.
- Conditions of early Earth:
  - Massive oceans present
  - Only small landmasses above surface of water
  - No ozone layer
  - Strong UV rays from Sun
  - Boiling oceans produced huge amounts of water vapour, making early earth extremely humid.
  - No free oxygen
  - Volcanic activity, heat, ash, dust and gases released into atmosphere
  - Atmosphere anoxic and full of toxic gases. Such as; water vapour (H<sub>2</sub>O), methane (CH<sub>4</sub>), ammonia (NH<sub>4</sub>), carbon dioxide (CO<sub>2</sub>), nitrogen (N<sub>2</sub>), hydrogen (H<sub>2</sub>).
    - Violent lightning storms common
- Description plausible since it matches with:
  - Conditions and chemicals on other planets.
  - The chemicals that erupt from volcanic vents today and probably always have.
- Chemicals of life contained within basic organic compounds:
  - Warter
  - Carbohydrates
  - Lipids
  - Proteins
  - Nucleic acids
- These compounds mainly made of carbon, hydrogen, oxygen, nitrogen (CHON).
- Elements needed to create basic organic compounds, already present in atmosphere.
- Lack of ozone layer, frequent electric storms and volcanic activity of early Earth could have provided energy for molecules to form.

#### Comparing Conditions of Early Earth to Today: Early Earth

	Early Earth	Today
Atmosphere	No ozone layer, UV radiation H <sub>2</sub> , CO <sub>2</sub> , hydrogen cyanide, N <sub>2</sub> , ammonia, water vapour No free oxygen	Ozone layer  Ozone layer
Oceans	<ul> <li>Plenty of oceans</li> <li>Boiling oceans producing water vapour</li> </ul>	<ul> <li>Constant 0° – 360°C (average)</li> </ul>
Temperature	<ul> <li>Extremely hot and humid (water vapour)</li> <li>1800°C</li> </ul>	Average temperature is 20°C     Temperate
Weather	<ul> <li>Huge and constant electrical storms</li> </ul>	<ul> <li>Fairly constant, occasional storms</li> </ul>
Geothermic Activity	Huge amounts of volcanic activity     Spewing ash and gas into atmosphere     Forming land	<ul> <li>Occasional volcanic eruptions</li> </ul>

Discuss the implications of the existence of arganic molecules in the cosmos for the origin of life on Earth (1.2.2)

- o For life to have originated, the following events need to have happened:
  - Required chemicals need to have been formed
  - Chemicals need to have come together in a self-replicating body
  - This body would need to have a form of protection for its contents.
  - It had to be able to use an energy source to replicate itself
  - Organic molecules needed for life would have had to be present
  - These organic molecules could have been formed here on Earth, or been sent to Earth from outer space (the cosmos)

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