Acids and Bases

Questi	on 1:	Choose the correct answer for	each of t	he follow	ring questions:				
1.1	The element present in all acids								
	A	oxygen		carbon dioxide					
	C	nitrogen	D	hydre					
1.2	A sample of an acid in a beaker has a pH of 4. An alka. *o it. W.								
	the liqu	aid in the beaker?							
	A	\ decreases		stays the same					
	C	Increases	D	neutral					
1.3	Choose the incorrect combination:								
	A	HNH₄OH – Nitric acid.	В	H ₂ SO	sulphuric ac				
	C	HCl - Hydrochloric acid.	D		rbonic acid.				
1.4	The colour of blue litmus paper in an acid is								
	A	yellow	В	en					
	C	blue							
1.5	Which:	ochloric a and lithium?							
	A	water		hydrogen 8-					
	C	oxygen gas		on dioxide gas					
1.6	Which:	substance is always produ in	the rea.		n hydrochloric acid and sodium				
	hydroxi	ide?							
	A	water		hydrogen gas					
	C	oxygen gas		ecipitate					
1.7	Which formula repr								
	А КОН			KCI					
	C	CH3OH		CH₃COOH					
1.8	T	od of adding an a	irop b	y drop:					
		neutralisation	В	precipita	ation				
		eva-	D	decomp	osition	[8]			
Quest.		.ne genera reactio	n for an	acid with	the following compounds:				
2.1	Base:								
2.2	Carbonate:								
2.3	Metal: Street victorial					7-3			

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