

(iii) Suggest why the mean bond enthalpy of a C=C bond is less than twice the mean bond enthalpy of a C—C bond.

(1)

(iv) Use the mean bond enthalpy data in the table, and the equation given below, to calculate a value for the standard enthalpy change of combustion of propene.

(3)

Bond	Mean bond enthalpy / kJ mol <sup>-1</sup>
C=C	612
C—C	347
C—H	413
O=O	498
C=O	805
O—H	464



Answer = \_\_\_\_\_ kJ mol<sup>-1</sup>



# Gce Edexcel Chemistry 2015 January Papers

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## **Table of Contents Gce Edexcel Chemistry 2015 January Papers**

1. Understanding the eBook Gce Edexcel Chemistry 2015 January Papers
  - The Rise of Digital Reading Gce Edexcel Chemistry 2015 January Papers
  - Advantages of eBooks Over Traditional Books
2. Identifying Gce Edexcel Chemistry 2015 January Papers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Gce Edexcel Chemistry 2015 January Papers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Gce Edexcel Chemistry 2015 January Papers
  - Personalized Recommendations
  - Gce Edexcel Chemistry 2015 January Papers User Reviews and Ratings
  - Gce Edexcel Chemistry 2015 January Papers and Bestseller Lists
5. Accessing Gce Edexcel Chemistry 2015 January Papers Free and Paid eBooks
  - Gce Edexcel Chemistry 2015 January Papers Public Domain eBooks
  - Gce Edexcel Chemistry 2015 January Papers eBook Subscription Services

- Gce Edexcel Chemistry 2015 January Papers Budget-Friendly Options
- 6. Navigating Gce Edexcel Chemistry 2015 January Papers eBook Formats
  - ePub, PDF, MOBI, and More
  - Gce Edexcel Chemistry 2015 January Papers Compatibility with Devices
  - Gce Edexcel Chemistry 2015 January Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Gce Edexcel Chemistry 2015 January Papers
  - Highlighting and Note-Taking Gce Edexcel Chemistry 2015 January Papers
  - Interactive Elements Gce Edexcel Chemistry 2015 January Papers
- 8. Staying Engaged with Gce Edexcel Chemistry 2015 January Papers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Gce Edexcel Chemistry 2015 January Papers
- 9. Balancing eBooks and Physical Books Gce Edexcel Chemistry 2015 January Papers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Gce Edexcel Chemistry 2015 January Papers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Gce Edexcel Chemistry 2015 January Papers
  - Setting Reading Goals Gce Edexcel Chemistry 2015 January Papers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gce Edexcel Chemistry 2015 January Papers
  - Fact-Checking eBook Content of Gce Edexcel Chemistry 2015 January Papers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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