# NOMENCLATURE PACKET

Worksheet I: Binary Ionic Compounds (representative metals) - metals from groups 1A, 2A, and 3A (1, 2, and 13) have constant charges as ions and do NOT get Roman Numerals in their names

#### 1. Name the following ionic compounds:

| α.  | Al <sub>2</sub> O <sub>3</sub> | Aluminum oxide     |  |  |
|-----|--------------------------------|--------------------|--|--|
| ь.  | Cs2O                           | Cesium oxide       |  |  |
| c.  | Rb <sub>3</sub> N              | Rubidium nitride   |  |  |
| d.  | Ca <sub>3</sub> N <sub>2</sub> | Calcium nitride    |  |  |
| e.  | SrSe                           | Strontium selenide |  |  |
| f.  | Cs₂S                           | Cesium sulfide     |  |  |
| 9.  | Al <sub>2</sub> S <sub>3</sub> | Aluminum sulfide   |  |  |
| h.  | LiBr                           | Lithium bromide    |  |  |
| i.  | Mg <sub>3</sub> N <sub>2</sub> | Magnesium nitride  |  |  |
| 11. | CaF <sub>2</sub>               | Calcium fluoride   |  |  |

# 2. Write the chemical formula for the following ionic compounds:

| a.         | Barium nitride          | Ba*2             | N-3  | Ba <sub>3</sub> N <sub>2</sub> |
|------------|-------------------------|------------------|------|--------------------------------|
| ь.         | Indium Fluoride         | In-3             | F-1  | InF <sub>3</sub>               |
| <b>c</b> . | Calcium oxide           | Ca-2             | 0.5  | CaO                            |
| d.         | Sodium nitride          | Na <sup>-1</sup> | N-3  | Na <sub>3</sub> N              |
| e.         | Magnesium chloride      | Mg*2             | CI-1 | MgCl <sub>2</sub>              |
| f.         | Potassium oxide         | K*1              | 0.2  | K <sub>2</sub> O               |
| 9.         | Magnesium oxide         | Mg*2             | 0-2  | MgO                            |
| h.         | Potassium sulfide       | K*1              | 5-2  | K25                            |
| i          | Lithium nitride         | Li               | N-3  | Li <sub>3</sub> N              |
| J-         | Strontium fluoride      | 5r*2             | F-1  | SrF2                           |
| k.         | Aluminum sulfide        | A1-3             | 5-2  | Al <sub>2</sub> S <sub>3</sub> |
| L          | Duplicate problem - rem | oved             |      |                                |

# **Ions And Their Chemical Formula Answers**

V. K. Sally, Chhaya Srivastava, Goyal Brothers Prakashan

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