

IONIC BONDING 1

Name: _____

Ionic bonding occurs between a metal and a non-metal. In the chart below show the transfer of electrons with an arrow.



1. $\text{K} + \text{O}$	$\text{K} \cdot + \cdot \ddot{\text{O}} \cdot \rightarrow \text{K}$	K_2O
2. $\text{Mg} + \text{O}$	$\text{Mg} \cdot + \cdot \ddot{\text{O}} \cdot \rightarrow$	MgO
3. $\text{Ca} + \text{S}$	$\text{Ca} \cdot + \cdot \ddot{\text{S}} \cdot \rightarrow$	CaS
4. $\text{Na} + \text{Br}$	$\text{Na} \cdot + \cdot \ddot{\text{Br}} \cdot \rightarrow$	NaBr
5. $\text{H} + \text{Cl}$	$\text{H} \cdot + \cdot \ddot{\text{Cl}} \cdot \rightarrow$	HCl
6. $\text{Al} + \text{Cl}$	$\cdot \ddot{\text{Cl}} \cdot + \text{Al} \cdot + \cdot \ddot{\text{Cl}} \cdot \rightarrow$	AlCl_3
7. $\text{N} + \text{O}$	$\cdot \ddot{\text{O}} \cdot + \cdot \ddot{\text{N}} \cdot \rightarrow$	N_2O_3
8. $\text{Cr}^{+3} + \text{S}^{2-}$	$\cdot \ddot{\text{S}} \cdot + \text{Cr} \cdot + \cdot \ddot{\text{S}} \cdot \rightarrow$	Cr_2S_3

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