

Chemical Bonding Review

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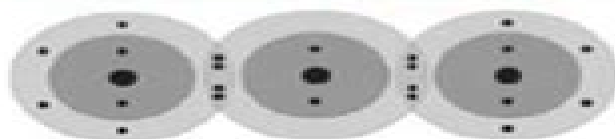
Name: _____

Directions: Answer each of the following questions by circling the correct answer choice.

- Which two elements have the same number of valence electrons?
A. Boron and Aluminum
B. Carbon and Nitrogen
C. Argon and Helium
D. Silicon and Aluminum
- What type of bond forms when electrons are transferred?
A. Covalent
B. Ionic
C. Metallic
D. Nonmetallic
- What is the chemical symbol for a potassium ion?
A. K^{+1}
B. K^{-1}
C. P^{+1}
D. P^{-1}
- How many electrons are needed for a filled outermost energy level in the Lewis-dot structure below?



- A. 4
B. 0
C. 2
D. 1
- When atoms gain electrons during chemical bonding, they become
A. Positively charged ions
B. Neutral ions
C. Negatively charged ions
D. Chemically charged ions
 - Potassium chloride is an example of what type of bond?
A. Metallic bond
B. Colloid bond
C. Covalent bond
D. Ionic bond
 - The diagram below shows what type of bond?



- A. Ionic
B. Covalent
C. Metallic
D. Solution

Ionic And Metallic Bonding Test B Answers

Prof. Yosry Moustafa



Ionic And Metallic Bonding Test B Answers:

NEET NATIONAL ELIGIBILITY CUM ENTRANCE TEST CHEMISTRY CLASS 11 VOLUME I Priti Singhal, 2024-11-12 The content in this book has been meticulously aligned with the latest NEET syllabus and trends reflecting insights from previous years papers We have focused on high yield topics and included chapter summaries topic wise MCQs and mock tests to provide a structured approach to learning Whether it s mastering Biology tackling numerical in Physics or understanding the complexities of Organic Chemistry this book offers step by step solutions to help students at every stage of their preparation

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