

Electron Configuration of Ions

Write out the configuration of each ion on the line. Then use the key in the right column to match the configuration to the noble gas that has the same configuration. Most of the noble gases will be used more than once. Some may not be used at all.

1. Na^+ : _____2. F^- : _____3. K^+ : _____4. Sr^{2+} : _____5. Mg^{2+} : _____6. O^{2-} : _____7. Br^- : _____8. P^{3-} : _____9. Li^+ : _____10. Fe^{2+} : _____11. S^{2-} : _____12. Ca^{2+} : _____13. Cl^- : _____14. Zn^{2+} : _____15. Sc^{3+} : _____

A. Helium (He)

B. Neon (Ne)

C. Argon (Ar)

D. Krypton (Kr)

E. Xenon (Xe)

F. Radon (Rn)

G. Does not match a noble gas.

He

Ne

Ar

Kr

Xe

Rn

Ion Electron Configuration Worksheet 2014

M Tight



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