

1. The use of a color-matching kit off the ground is another technique of identifying the color.
2. When the color-matching kit is used, a flame test tube is left off the ground.
3. When the color-matching kit is used, a flame test tube is left off the ground.
4. When the color-matching kit is used, a flame test tube is left off the ground.
5. When the color-matching kit is used, a flame test tube is left off the ground.
6. When the color-matching kit is used, a flame test tube is left off the ground.
7. When the color-matching kit is used, a flame test tube is left off the ground.
8. When the color-matching kit is used, a flame test tube is left off the ground.
9. When the color-matching kit is used, a flame test tube is left off the ground.
10. When the color-matching kit is used, a flame test tube is left off the ground.

CONCLUSION

Write about your learning about the properties of the elements tested.

When did you see in the experiment? If something worked well, you should have noticed that the sample flame originally burned with an orange-yellow color. When the color-matching kit was placed in the flame, a thermal with a brighter yellow color is burned with an orange color.

When the color-matching kit was placed in the flame, a thermal with a brighter yellow color is burned with an orange color. When the color-matching kit was placed in the flame, a thermal with a brighter yellow color is burned with an orange color. When the color-matching kit was placed in the flame, a thermal with a brighter yellow color is burned with an orange color.

think about this



None, or the same of the meaning of this book, accounts have identified 120 basic elements in chemical elements, but not so many that they can be changed into time.

In chemistry, the elements are identified by their chemical properties, but it is a possibility.

ON YOUR OWN

1. Write of the following elements, period, group, and block in the periodic table of the elements.
2. Explain the difference between an atom and an element.

THE PERIODIC TABLE OF THE ELEMENTS

The periodic table of the elements (Figure 8.1) is a chemical's most important tool. It is more than just a listing of the elements. You see, as a chemist learned more about the behavior and properties of elements, they discovered they could arrange them based on what they were made of and how they behaved. Figure 8.1 shows that, instead of the table that you have seen, it is a table of the elements. Figure 8.1 shows that, instead of the table that you have seen, it is a table of the elements. Figure 8.1 shows that, instead of the table that you have seen, it is a table of the elements.

FIGURE 8.1
The Periodic Table of the Elements

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Mike Jess



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