

Please write clearly in block capitals.

Centre number

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Candidate number

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Surname

\_\_\_\_\_

Forename(s)

\_\_\_\_\_

Candidate signature

\_\_\_\_\_

# GCSE Chemistry

# H

Higher Tier Unit Chemistry C3

Wednesday 15 June 2016

Afternoon

Time allowed: 1 hour

## Materials

For this paper you must have:

- a ruler
- the Chemistry Data Sheet (enclosed).

You may use a calculator.

## Instructions

- Use black ink or black ball-point pen.
- Fill in the boxes at the top of this page.
- Answer **all** questions.
- You must answer the questions in the spaces provided. Do not write outside the box around each page or on blank pages.
- Do all rough work in this book. Cross through any work you do not want to be marked.

## Information

- The marks for questions are shown in brackets.
- The maximum mark for this paper is 60.
- You are expected to use a calculator where appropriate.
- You are reminded of the need for good English and clear presentation in your answers.
- Question 2 should be answered in continuous prose.  
In this question you will be marked on your ability to:
  - use good English
  - organise information clearly
  - use specialist vocabulary where appropriate.

## Advice

- In all calculations, show clearly how you work out your answer.



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**L Towne**



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choose 1 answer the solution becomes negatively charged due to the majority  $Cl^-$  ions a the solution becomes negatively charged due to the majority  $Cl^-$  ions the solution becomes positively charged due to the stronger  $Mg^{2+}$  ions b the solution becomes positively charged due to

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outside express your answer in terms of the total charge  $q$  on the surface  $p = z \cos \theta$   $rsin \theta$   $z \cos \theta$   $q$  figure 2.4 problem 2.7  
consider a slice of the shell centered on the  $z$  axis see

