



IONIC AND METALLIC BONDING

Practice Problems

Do your best to solve these problems on your own.

SECTION 7.1 IONS

1. How many electrons does each element of atoms have? How many protons does each have? How many neutrons does each have? (See Table 7.1.)

a. Fe

b. Li

c. Br

2. How many valence electrons does each element have? (See Table 7.1.)

a. Fe

b. Br

c. Br

3. How many valence electrons does each element have? (See Table 7.1.)

a. Fe

b. Br

c. Br

d. Fe

e. Br

4. How many valence electrons does each element have? (See Table 7.1.) How many protons does each element have? (See Table 7.1.)

5. How many valence electrons does each element have? (See Table 7.1.) How many protons does each element have? (See Table 7.1.)

a. Fe

b. Br

c. Br

d. Fe

e. Br

f. Br

6. How many valence electrons does each element have? (See Table 7.1.)

a. Fe

b. Br

c. Fe

d. Br

7. How many valence electrons does each element have? (See Table 7.1.)

a. Fe

b. Br

c. Br

d. Br

8. How many valence electrons does each element have? (See Table 7.1.)

a. Fe

b. Br

c. Br

d. Fe

e. Br

f. Br

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Xiang Xie



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