Name: Date:		Per:
Ionic Bonding WebQu		
Part 1: Go to http://www.visionleaming.com/library/module	, yilenye	.php?mid=66
head the Chemical Bonding and lonic Bonding sections	and a	namer the tollowing
questions:		
Elements react with each other to form		
Elements react with each other to form What two elements is table salt composed of?		and
Elements react with each other to form What two elements is table salt composed of? Who contributed to our knowledge of valence electrons		and
Elements react with each other to form What two elements is table salt composed of?		and

- Look at the picture of the NaCl Crystal Schematic. Explain why there is no such thing as a single molecule of salt:
- Click the link The reaction of actions with chloring and watch through the video there.
 Briefly describe each of the steps involved with the reaction of sodium with chlorine.

Part 2: Watch the following clips on lonic Bonding:

- http://watch?video.net/lonic-bondmov-video-pRGtdkaGwKP.html
- http://www.youtube.com/leatch?ps/CojcCycV/www.rkynove.the.cov/eleot.part/.
- http://www.mhine.com/physics/chemistry/enimations/chang. Ze. espition Lo2: 11 avil (you can click on the NAF lonic compound and see an enimation on how it forms!)

Part 3: Explore the link below. You amplies on the elements and compounds in the video to each into high high Woontent Exponents a pomisher sister. 10/pombers/media/048, electro, bonding, swi

B. Describe how ionic compounds form:

Part 4: Take the quiz at the following website. Record your score here: of 10. http://www.softschools.com/guizzes/shemistryfonic_bonding_formation/guiz299.html

Part 5: Dissociation of Salt (an ionic compound). Watch the following videos:

- http://www.voutube.com/watch?vvEBtGcTAJE4p
- http://www.mhhe.com/physaci/chemistry/essentialchemistry/liash/molyle1.swf
- Go to http://phat.colorado.edu/en/almulation/augar-and-aaff-actutiona.
 Use the simulation and explain how salt and sugar disactive in water.

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JR Anderson

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