

- Directions:**
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.
 2. An atom with a positive charge is a cation.
 3. An atom with a negative charge is an anion.
 4. The number of protons in an atom is the atomic number.

Ion Practice Set

1. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.
 2. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.

Complete the following table, using the periodic table in the back of your book.

	Element Name	Ion Symbol	Number of Protons	Number of Electrons	Charge on Ion (lost or gained)	Ion Name
1	Sodium	Na^+	11	10	lost one	cation
2	Chlorine	Cl^-	17	18	gained one	anion
3	Strontium	Sr^{2+}	38	36	lost two	cation
4	Phosphorus	P^{3-}	15	18	gained three	anion
5	Calcium	Ca^{2+}	20	18	lost two	cation
6	Chromium	Cr^{3+}	24	21	lost three	cation
7	Hydrogen	H^+	1	0	lost one	cation
8	Lithium	Li^+	3	2	lost one	cation
9	Fluorine	F^-	9	10	gained one	anion
10	Magnesium	Mg^{2+}	12	10	lost two	cation
11	Sulfur	S^{2-}	16	18	gained two	anion
12	Aluminum	Al^{3+}	13	10	lost three	cation
13	Oxygen	O^{2-}	8	10	gained two	anion
14	Boron	B^{3+}	5	2	lost three	cation
15	Carbon	C^{4-}	6	10	gained four	anion
16	Nitrogen	N^{3-}	7	10	gained three	anion
17	Helium	He^0	2	2	lost none	neutral
18	Neon	Ne^0	10	10	lost none	neutral
19	Silver	Ag^+	47	46	lost one	cation
20	Gold	Au^3+	79	76	lost three	cation

Ion Practice Set Key

Michael Seilmaier

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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Table of Contents Ion Practice Set Key

1. Understanding the eBook Ion Practice Set Key
 - The Rise of Digital Reading Ion Practice Set Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Ion Practice Set Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ion Practice Set Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ion Practice Set Key
 - Personalized Recommendations
 - Ion Practice Set Key User Reviews and Ratings
 - Ion Practice Set Key and Bestseller Lists
5. Accessing Ion Practice Set Key Free and Paid eBooks

- Ion Practice Set Key Public Domain eBooks
- Ion Practice Set Key eBook Subscription Services
- Ion Practice Set Key Budget-Friendly Options
- 6. Navigating Ion Practice Set Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Ion Practice Set Key Compatibility with Devices
 - Ion Practice Set Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ion Practice Set Key
 - Highlighting and Note-Taking Ion Practice Set Key
 - Interactive Elements Ion Practice Set Key
- 8. Staying Engaged with Ion Practice Set Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ion Practice Set Key
- 9. Balancing eBooks and Physical Books Ion Practice Set Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ion Practice Set Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ion Practice Set Key
 - Setting Reading Goals Ion Practice Set Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ion Practice Set Key
 - Fact-Checking eBook Content of Ion Practice Set Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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