Reminders:

 In a neutral atom the number of protons equals the number of electrons.

An atom can
 NEVER gain or lose
 protons
 The number of

 The number of protons equals the atomic number

Ion Practice Set

1. What is an ion?

Es un átomo o un grupo de átomos (molécula) con una carga negativa o positiva

2. What does the number next to the ions signify?

La carga que llevan

Complete the following table, using the periodic table in the back of your book.

ex	ELEMENT NAME Fluorine	ION SYMBOL	NUMBER OF PROTONS	NUMBER OF ELECTRONS	NUMBER OF ELECTRONS LOST OR GAINED gained one
2	Azufre	S-2	16	18	gained two
3	potassium	K*	19	18	lost one
4	Calcio	Ca ⁺²	20	18	Lost two
5	Bromo	Br	35	36	Gained one
6	Estroncio	Sr ⁺²	38	36	Lost two
7	Hidrogeno	н*	1	2	Gained one
8	Oxigeno	0.2	8	10	gained two
9	Magnesio	Mg*2	12	10	lost two
10	aluminum	Al*2	13	10	Lost two
11	Selenio	Se ⁻²	34	36	Gained two
12	Hidrogeno	н*	1	o	Lost one
13	lithium	Li*	3	2	lost one
14	Rubidio	Rb⁻	37	38	Gained one
15	Cloro	CI:	17	18	Gained one

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