

6(a)(ii)	$\frac{492.3}{688.3} \times 100 \quad \text{OR} \quad \frac{492}{688} \times 100$ $= 71.5\%$	2	1 mark for both <i>M</i> correctly placed.
6(b)	$3\text{Ca}(\text{OH})_2 + 2\text{H}_3\text{PO}_4 \longrightarrow \text{Ca}_3(\text{PO}_4)_2 + 6\text{H}_2\text{O}$	1	Allow multiples.
6(c)	$\begin{array}{c} \text{Ca} \\ 1.67 \\ 40.1 \\ \\ = 0.042 \\ 1 \end{array}$ $\begin{array}{c} \text{H} \\ 0.17 \\ 1 \\ \\ 0.17 \\ 4 \end{array}$ $\begin{array}{c} \text{P} \\ 2.99 \\ 31 \\ \\ 0.084 \\ 2 \end{array}$ $\begin{array}{c} \text{O} \\ 5.33 \\ 16 \\ \\ 0.333 \\ 8 \end{array}$ $\text{CaH}_2\text{P}_2\text{O}_8$ OR $\text{Ca}(\text{H}_2\text{PO}_4)_2$ OR $x = 2$ Alternative $\begin{array}{c} \text{Ca} \\ 1.67 \\ 40.1 \\ \\ = 0.042 \\ 1 \end{array} \quad \begin{array}{c} \text{H}_3\text{PO}_4 \\ 8.09 \\ 97.0 \\ \\ 0.083 \\ 2 \end{array}$ $\text{CaH}_2\text{P}_2\text{O}_8$ OR $\text{Ca}(\text{H}_2\text{PO}_4)_2$ OR $x = 2$	1 1 1 1 If $x=2$ with no working, allow M4 only. Ca = 1.67 g (M1). Mark for dividing by correct <i>A_r</i> in Ca and P (M2). If M1 incorrect can only score M2 Correct ratio (M3). Value of <i>x</i> or correct formula (M4). Ca = 1.67 g (M1). Mark for dividing by correct <i>A_r</i> / <i>M_r</i> in Ca and H₃PO₄ (M2). If M1 incorrect can only score M2 Correct ratio (M3). Value of <i>x</i> or correct formula (M4).	

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