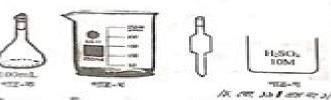
संभावत् । सहस्र करः, ५ - 20 हः 1107 अदः कार्यास्त्र करः, 51 - 36.5 हः/कलः सर्वास्त्र व्यक्तित् , ५ - 11. सर्वास्त्र व्यक्तित् , ५ - 11.

ত্রতীনতে বলিও X-ক্ষান্তিতে 10 mg এব মেছে কম পরিমাণ হব X-ক্ষান্তিতে $0.5 \times 2\%$ $\left(\frac{W}{V}\right)$ 16.7 কমুনা হৈছে কাম করা ছাছেছে। স্থান্ত X-ক্ষান্তিতি হলো সোত্রে ক্ষান্তান্তিতি বালা ক্ষান্তিত ক্ষান্তিতি হলো মাজে ক্ষান্তান্তিতি বালা ক্ষান্তিত ক্ষান্তিতি হলো মাজে ক্ষান্তান্তিতি বালা ক্ষান্তিত তি হলা ম্বান্তিত ক্ষান্তান্তিতি বালা ক্ষান্তিত ক্ষা

হাতি হৈছে পাণ্ডতি আমার্যনিক বুগগত বিজেকবার একটি পাণ্ডতি।
এই বিজেপতি বসায়নে মল পরিমাণ আমার্যনিক উপাদনে সমান্তবন্তার
এই বিজেপতি বসায়নে মল পরিমাণ আমার্যনিক উপাদনে সমান্তবন্তার
এই কান্তবিজ্ঞান বা 2-dnal, তারন প্রবা) আমার্যনিক শানার্য নিজে
এইক কান্তবা হা। কান্তবা প্রবাহিত প্রবাহিত কান আমান।
এইক কান্তবিজ্ঞান প্রবাহিত আমার্যনিক পরিমাণ্ড
এইক ক্ষেম্বিক মার্যনিক কান্তবার কান্তবার মার্যনিক স্বান্তবিজ্ঞান স্বান্তবিজ

প্রথার বেটানা কোনো কেতে কম জুকির রাসায়নিক প্রবা বাবস্থুত বয়। কাল পরিবেশ দুখাবের সম্ভাবনা কমে আলে। জাই সেমি-মাইজে পশ্বতি প্রতিবেশ বাশ্বর।

EECJES



- ত, তাইজার প্রথক কীপু
- সেমি-মাইক্রো পশ্বতি পরিবেশ বান্ধর কেন?
- লংগটির জন্য প্রযোজ্য ছ্যাজার্চ প্রতীকটি অভবন করে। এবং ব্যাখ্যা করে।
- উলিপকের আলোকে দ্রবদ থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার দ্রবদ তৈরি করার প্রক্রিয়া ব্যাব্যা করে। ৪

৩ নং প্রয়োর উত্তর

থিমেবণীয় নিস্তির বীমের উপর রাইভার স্থাপন করনে বীমের প্রতি মধ্যকের জন্য যে তর পাওয়া যায়, তাকে রাইভার ধ্রুবক বলে।

তি তেনি ইতেন পংশতি রাসারনিক গুণগত বিশ্লেষণের একটি পংশতি।
টি হৈছেদির রসায়নে ছল পরিমাণ রাসায়নিক উপাদান সনাক্তরণ ও
পনিবাহ হিংহালে ব্যবস্ত ছল। এ পংশতিতে দুব অল পরিমাণ

০.০০ হৈতে ০.2g করিন বা 2.4ml. তরল লবা) লাবায়নিক পদার্থ নিয়ে
প্রীয়া করা হয়। মতে পরিবেশ দুখণ উল্লেখযোগ্যভাবে কমে আমে।

ইবছ, কোনো কোনো কেন্তে কম খুঁকির রাসায়নিক লবা ব্যবস্থৃত হয়।

ইতং পরিবেশ দুখণের সন্ধাবনা কমে আসে। ভাই সেমি-মাইন্ডো পংশ্বিত
পরিবেশ ব্যক্তর।

টি ইন্দাকের প্রবণটি হলো সালফিউরিক এশিভ (H,SO₄) এর 10 M ইল। H,SO₄ একটি জারক ও জয়কারী পদার্থ। H₂SO₄ এর 10 M ইল হলে পাচ H₂SO₄ প্রবণ। গাচ H₂SO₄ পদার্থকে জয়কারী ছাজার্ত ইতিবে অন্তর্ভন্ত করা হয়। গাচ H₂SO₄ এর জন্য জয়কারী ছালার্ত ইতিবে অন্তর্ভন্ত করা হয়। গাচ H₂SO₄ এর জন্য জয়কারী ছালার্ত ইতিবিটে জনকন করে ব্যাখ্যা করা হলো:

weeds (Corresive)
ক্ষরবাধী রাম্মান্ত্রিক উপাধান বলতে সে সকল রাম্মান্ত্রিক প্রবাধে বুরার যার সংস্পার্ক তিক ও জনানা পালার্ক্তর মানাক্রক ক্ষতি সাধিত বহা। এতে তুক পুচে
THE WHEN SERVICE CHEST RESOLUTION TO THE STATE OF THE SERVICE
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ভিন্নীপাৰের প্রকাটি হলে 1606 H₂SO, প্রকা একা এর বেচর 1M ঘনমানের প্রকা গাড়ুত করতে যান ১ম ও এর কাডবর বাধারে করতে মধে।

১ম কাজধারটি বাংলা 100 ml. আহাতনমিভিক চ্যাম্ব ও ওর টি বাংলা সিংগটি। বেংমতু আহাতনমিভিক চ্যাম্বের আহাতন 100 ml., তাই আমানের 100 ml. 1M অন্যাত্রার H_SSO, প্রবণ তৈরি করতে হবে। প্রবংগর অমুকরণ সূত্র যতে আমরা তারি,

$$V_1S_1 = V_2S_2$$

 $V_1 = \frac{V_2S_2}{S_1}$
 $= \left(\frac{100 \times 1}{10}\right) \text{ mL}$
 $= 10 \text{ mL}$

এখনে, আম্মিক জন্তন, V, -? মনমাত্রা, S, = 10M শেষ জন্তন, V₂ = 100 mL মনমাত্রা, S₂ = 1M

खाउन्दर, $10mL\ H_0SO_4$ सदरशंद ऋषण शाबि (यांच कराव शाव (100-10) $mL=90\ mL$

আবার, শিপেটের সংঘাতে 10 ml. প্রমত খনমাত্রার তবল নিয়ে তার সাবে আয়তন্ত্রিতিক ফ্রান্ডে 90 ml. পানি মোণ করলে আন্তর্ভর কাজিনত খনমাত্রার প্রবণ পাওয়া যাবে :

运用机工







SE-A

B-2 - 13

(R. (M. 20 | 20 M to

- ক, আন্টিঅভিডেউ কাকে বলেঃ
- সামা ধ্রক K, এর মান কথনো অসীম হয় না–ব্যাখ্যা করে।
- প্ 🔥 ম ভিত্তের মন্ত্রটিকে উত্তপ্ত করার কৌশল বর্ণনা করে। 🕠
- ঘ, ৪ ও ে ডিছের সাথে সম্পর্কিত রাসায়নিক প্রবাদির ব্যবহারকালে দ্বাস্থ্য কৃতি হাসে করণীয় উপায় সম্পর্কে বিশেষণ করে।

৪ মং প্রয়োর উভর

ত্র আাণ্টিঅভিডেন্ট এক ধরনের রাসারনিক বা কৃত্রিম খালা সংক্রক যা মুখত খালা উপাদানসমূহকে জারণ প্রতিয়ায় কার্ম থেকে রক্ষা করে যাবার সংক্রকণ করে।

া সাম্যাচক K, এর মান কখনো অশীম বা পূন্য হতে পারে মা। ডাঙ্কিটি ব্যাখ্যা করার জন্য নিচের উভমুখী সমীকরণটি বিবেচনা করি— A + B ,—— C + D

Hsc Chemistry 1 First Chapter Notes

J Ma

Hsc Chemistry 1 First Chapter Notes:

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