

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

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Whispering the Secrets of Language: An Emotional Journey through **Holt Modern Chemistry Homework Answers Reviewing Concepts**

In a digitally-driven world where screens reign great and instant connection drowns out the subtleties of language, the profound secrets and psychological nuances hidden within phrases frequently move unheard. Yet, nestled within the pages of **Holt Modern Chemistry Homework Answers Reviewing Concepts** a interesting fictional value pulsating with fresh thoughts, lies a fantastic journey waiting to be undertaken. Published by a talented wordsmith, this charming opus invites viewers on an introspective journey, lightly unraveling the veiled truths and profound affect resonating within the material of each and every word. Within the mental depths of the touching evaluation, we will embark upon a heartfelt exploration of the book is core subjects, dissect their interesting publishing design, and succumb to the powerful resonance it evokes serious within the recesses of readers hearts.

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