Periodic Table: Practice Test

1	Protons have a negative charge.
2.	The modern periodic table is organized according to atomic mass.
3.	The horizontal rows in the periodic table are known as groups.
4.	The elements in a group of the periodic table have different characteristics.
5.	Describing a metal as ductile means that it can be pounded into a new shape.
6.	The most chemically reactive metals are in Group 8 of the periodic table.
7.	In general, the physical properties of nonmetals are different from to the properties of metals.
8.	Atoms of the halogen family of elements typically gain two electrons when they react.
9.	If you know the Atomic number, you can figure out the number of protons and neutrons.
10.	Positively charged particles in an atom's nucleus are called <u>neutrons</u> .
11.	The valence electrons are those electrons in the outermost shell.
12.	Each family in the periodic table has its own characteristic properties based upon its number of valence electrons.
13.	The mass number is the protons plus electrons.
14.	Thomson's model introduced the <u>nucleus</u> .
15.	The first model to introduce energy levels was Rutherford.
16.	The modern day atom model is known as the Bohr model.
17.	Isotopes of the same element have the same number of protons.
18.	Isotopes of the same element have a different atomic number.
19.	An element has an atomic number of 12. That means it has 24 protons.
20	If an element has an atomic number of 10 and a mass number of 21, the element has 10 neutrons.

Completion

Complete each statement.

- 21. The ______ is the very small center core of an atom.
- 22. The property of an element that indicates the number of protons in its atoms is the ______
- Mendeleev discovered that periodic patterns appeared when he arranged the elements in order of atomic mass.

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