- g) A solution has pH 4.2596. Find out hydrogen ion concentration.
- h) Define the terms
  - i) gangue
  - ii) ore
- i) State two purposes of calcination.
- j) Write composition of duralumin.
- k) What is tack property of rubber?
- Give two uses of thermocole based on its two different properties.

## Attempt any <u>FOUR</u> of the following:

- a) Distinguish between atomic number and atomic mass number.
- b) Write the electronic configuration of the following (in s, p, d, f orbitals) 2He<sup>4</sup>, 6C<sup>12</sup>, 12Mg<sup>24</sup>, 19K<sup>39</sup>.
- c) Define covalency. Describe the formation of Nitrogen

  Molecule.
- d) Define degree of ionization. Describe the factors affecting degree of ionization.
- e) Describe electro-refining process with sketch.
- f) A current of 2 amperes was passed through a solution of copper sulphate for 16 minutes and 5 seconds. Calculate the amount of copper deposited on cathode.

(At.wt. of Cu = 63.5)

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# **Msbte Chemistry Question Papers**

**Michael Brown** 

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