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web gabrielle m 9 years ago i m working on chemical reactions double and single replacement on flvs now my first question for this video is how do you have 4 aluminum atoms when it says 2Al subscript 2 do you just add the 2 and the subscript 2 same thing goes with the O_3 i am so lost

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web google classroom balance the following chemical equation $\text{Mg} + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$ note all reactants and products require a coefficient of at least one stuck review related articles videos or use a hint report a problem $7.41 \times 10^5 \text{ J}$ $\theta = \pi/8$ $5.2 \times 10^9 \text{ s}$

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web a common type of stoichiometric relationship is the mole ratio which relates the amounts in moles of any two substances in a chemical reaction we can write a mole ratio for a pair of substances by looking at the coefficients in front of each species in

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