

Experiment Problems 1

1. Complete each:

- a. Ag^+ OH^- aq
 b. Ag^+ OH^- aq
 c. Cl^- OH^- aq
 d. Cl^- OH^- aq
 e. Ag^+ Ag^+ aq
 f. Ag^+ Ag^+ aq
 g. Cl^- AgCl aq
 h. Cl^- AgCl aq

- i. Ag^+ AgCl aq
 j. Cl^- AgCl aq
 k. Cl^- AgCl aq
 l. Cl^- AgCl aq
 m. Ag^+ AgCl aq
 n. Ag^+ AgCl aq
 o. Cl^- AgCl aq
 p. Cl^- AgCl aq

2. Complete by writing chemical or ionic equations:



3. Complete molecular/total ionic equations, and separate ionic equations and net ionic:

| | Reactant | Product |
|----|---------------|---|
| a. | Ag^+ | |
| b. | | Ag^+ OH^- aq |
| c. | Cl^- | |

| | Reactant | Product |
|----|---------------|---------------|
| d. | Cl^- | |
| e. | Ag^+ | |
| f. | | Ag^+ |

4. Complete each equation. Chemical formulas are represented by subscripts and superscripts.



Making Connections With Factoring Hs Mathematics

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