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Chemical Equilibrium

Instructions:

- · Listen to an introduction from your recitation instructor.
- Work through the entire activity with your group. Each group member is responsible for contributing equally to this activity, and each person will turn in an assignment following the activity.
- No one is allowed to leave before the recitation period is over. If your group completes the
 activity early, use the extra time to help other groups.
- The recitation instructors will wrap up the session with an end-of-class 5-minute discussion.
- At the end of the class, one student from each table should gather the individual activity packets and turn them in to the recitation instructor as a set. Each activity packet should be complete.

| Concepts and Skills: | For Further Practice | |
|--------------------------|--------------------------|--|
| Chemical Equilibrium | Chapter 12, Section 12.4 | |
| The Equilibrium Constant | Chapter 12, Section 12.5 | |
| Le Chatelier's Principle | Chapter 12, Section 12.6 | |
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Padhraic Smyth

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