Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON METALS share electrons, attempting to attain a stable octet (8 outer electrons)in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

H_{2 brothrogen} is distornic

H + H → H → H-H Single Bond

2.) F2 thurston in discounts

3.) O2 oxygen is diatomic

O + O

 \bullet $\bigcirc =\bigcirc$

Double Bond

4.) No nitrogen is diatomic. Is this a triple bond?

N + N

5.) BF_{3 you need 3 fluorine atoms bere-}

 $\mathbf{R} + \mathbf{F}$

6.) Ammonia NH3 hist: how many hydrogen atoms are needed?

N + H

7.) Carbon dioxide CO₂

0 + c + 0

Methane CH₄ careful here 4 hydrogen atoms needed

C + H

9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!

H + O + H

10.) SO₂ hint: one pair of electrons from sulfur must be slit up for this one to work.

O + S + O

<u>Introduction To Chapter 9 Covalent Bonding Worksheet</u> <u>Answers</u>

Ralph Thomas Myers

Introduction To Chapter 9 Covalent Bonding Worksheet Answers:

Books in Print Supplement ,2002 Holt Chemistry Ralph Thomas Myers,2004

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