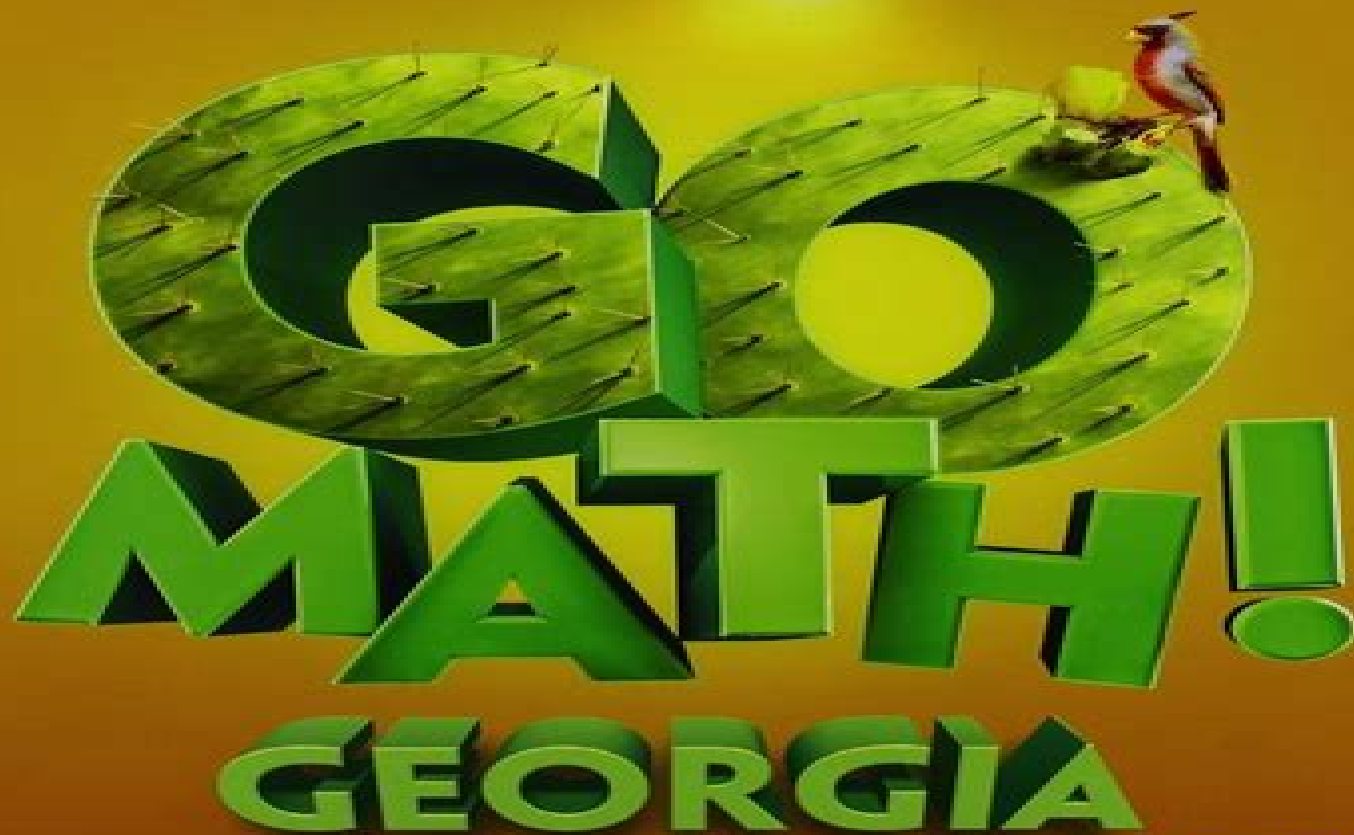


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 $2\text{ }3\text{Ca(OH)}_2 + 1\text{ }Al_2\text{SO}_4 \rightarrow 3\text{ }CaSO_4 + 2\text{ }Al(OH)_3$ type of reaction double displacement
 $3\text{ }3\text{Mg} + 1\text{ }Fe_2O_3 \rightarrow 2\text{ }Fe + 3\text{ }MgO$ type of reaction single displacement
 $4\text{ }1\text{ }C_2H_4 + 3\text{ }O_2 \rightarrow 2\text{ }CO_2 + 2\text{ }H_2O$

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 $NaOH + KNO_3 \rightarrow NaNO_3 + KOH$
 $CH_4 + O_2 \rightarrow CO_2 + H_2O$
 $2\text{ }Fe + 6\text{ }NaBr \rightarrow 2\text{ }FeBr_3 + 6\text{ }Na$
 $3\text{ }CaSO_4 + 6\text{ }NaOH \rightarrow 3\text{ }Ca(OH)_2 + 3\text{ }Na_2SO_4$
 $2\text{ }C_4H_8 + 12\text{ }O_2 \rightarrow 8\text{ }CO_2 + 8\text{ }H_2O$
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