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Gce Bible Study Past Question Papers

Jin-Ying Zhang



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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction $2\text{Ag}^+_{\text{aq}} + \text{Cr}_2\text{O}_7^{2-}_{\text{aq}} \rightarrow 2\text{Cr}_2\text{O}_7^{2-}_{\text{s}} + 2\text{Ag}^+_{\text{s}}$

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web jan 29 2020 the ionic form of this reaction is $2\text{Na}^+_{\text{aq}} + 2\text{Cl}^-_{\text{aq}} + \text{Cu}^{2+}_{\text{aq}} + \text{SO}_4^{2-}_{\text{aq}} \rightarrow 2\text{Na}^+_{\text{aq}} + \text{SO}_4^{2-}_{\text{aq}} + \text{CuCl}_2_{\text{s}}$ the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web these spectator ions ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{h}^+\text{aq} + \text{oh}^-\text{aq} \rightarrow \text{h}_2\text{ol}$ the spectator ions in koh^+aq and hno_3^-aq are k^+aq and no_3^-aq how to identify and list the

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