

100) Carbon-sulfur double bonds are not found in C_{60} molecules. What C_{60} molecule is a ring of single bonds. How many atoms of sulfur can there be in fullerenes of C_{60} molecules?

$$\text{Ans} = \text{Fullerenes } \text{C}_{60} \text{ has } 32 \text{ } \text{C}_2 \text{ units} \quad \text{C}_{60} \text{ has } 32 \text{ } \text{C}_2 \text{ units } \rightarrow 32 \times 2 = 64$$

$$x = \frac{64}{2} = 32$$

Answer is 32 100%

101) There only 3 nonbenzenoid forms in solid sulfur are not the Plausis.

102) Describe how can the Plausis forms arise.

Plausis forms arise from the reaction of sulfur with
carbon in the presence of a catalyst


in the form of a solid, as a result of the reaction
of sulfur with carbon 100%

103) Suggest other three can be other nonbenzenoid forms in solid sulfur.

the Plausis form, the form of sulfur
in the form of a solid, as a result of the reaction
of sulfur with carbon 100%

104) Sulfur dioxide is a compound of sulfur (atomic number 16) and oxygen.

Calculate the oxidation state of sulfur in SO_2 .

What is the oxidation number of sulfur in SO_2 ?

Ans = 4 100%



F321 May 2013 Chemistry

R Bogdan



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