



**PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2**

TIME: 1 HOUR

**VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).**

1. Write a scientific report which must include the following:

1.1 Aim

(1)

1.2 Apparatus and method

(5)

2. Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate pV for each set of results and record your answers in the table.

Volume (V) cm^3	Pressure (p) kPa		pV (J)

(6)

3. Analysis of the results:

3.1 Identify the following variables for the experiment:

- Independent
- Dependent
- Controlled

(3)

3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus $\frac{1}{p}$.

(18)

4. Conclusion

4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.

(1)

4.2 Write down a conclusion for this experiment.

(3)

5

Questions and discussion:

5.1 Explain why you have to wait for a few minutes before you can take readings.

(2)

5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.

(4)

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Table of Contents Memo Physics Grd11 2013

1. Understanding the eBook Memo Physics Grd11 2013
 - The Rise of Digital Reading Memo Physics Grd11 2013
 - Advantages of eBooks Over Traditional Books
2. Identifying Memo Physics Grd11 2013
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Memo Physics Grd11 2013
 - User-Friendly Interface
4. Exploring eBook Recommendations from Memo Physics Grd11 2013
 - Personalized Recommendations
 - Memo Physics Grd11 2013 User Reviews and Ratings
 - Memo Physics Grd11 2013 and Bestseller Lists
5. Accessing Memo Physics Grd11 2013 Free and Paid eBooks

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- Memo Physics Grd11 2013 eBook Subscription Services
- Memo Physics Grd11 2013 Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Memo Physics Grd11 2013 Compatibility with Devices
 - Memo Physics Grd11 2013 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Memo Physics Grd11 2013
 - Highlighting and Note-Taking Memo Physics Grd11 2013
 - Interactive Elements Memo Physics Grd11 2013
- 8. Staying Engaged with Memo Physics Grd11 2013
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Memo Physics Grd11 2013
- 9. Balancing eBooks and Physical Books Memo Physics Grd11 2013
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Memo Physics Grd11 2013
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Memo Physics Grd11 2013
 - Setting Reading Goals Memo Physics Grd11 2013
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Memo Physics Grd11 2013
 - Fact-Checking eBook Content of Memo Physics Grd11 2013
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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