

1

Determine the volume of CO_2 gas given off when excess MgCO_3 is added to 25.0 cm^3 of 0.400 mol/dm^3 HCl at room temperature and pressure.

C

✓



Use the following steps.

- Calculate the number of moles of HCl in 25.0 cm^3 of 0.400 mol/dm^3 of acid.

$$C = \frac{m}{V}$$

$$m = 0.4 \times 0.025$$

$$= 0.01$$

.....0.01..... mol

- Determine the number of moles of CO_2 gas given off.



$$0.01 \div 2 = 0.005$$

.....0.005..... mol

- Calculate the volume of CO_2 gas given off in cm^3 .

$$1\text{ mol} = 24\text{ dm}^3$$

$$0.005\text{ mol} = x\text{ dm}^3$$

$$x = 0.005 \times 24$$

$$= 0.12\text{ dm}^3$$

$$= 120\text{ cm}^3 (\times 1000)$$

.....120..... cm^3
[3]

Igcse Past Papers Questions On Stoichiometry

Clifford Lane



Igcse Past Papers Questions On Stoichiometry:

This book delves into Igcse Past Papers Questions On Stoichiometry. Igcse Past Papers Questions On Stoichiometry is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Igcse Past Papers Questions On Stoichiometry, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Igcse Past Papers Questions On Stoichiometry
- Chapter 2: Essential Elements of Igcse Past Papers Questions On Stoichiometry
- Chapter 3: Igcse Past Papers Questions On Stoichiometry in Everyday Life
- Chapter 4: Igcse Past Papers Questions On Stoichiometry in Specific Contexts
- Chapter 5: Conclusion

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