

## Chemistry Periodic Table Test Chapter 6 Chem IH

**Matching** Match each item with the correct statement below.

- |                      |                     |
|----------------------|---------------------|
| a. electronegativity | e. transition metal |
| b. ionization energy | f. periodic law     |
| c. electrons         | g. group            |
| d. metal             | h. period           |

- \_\_\_\_\_ 1. vertical column in the periodic table
- \_\_\_\_\_ 2. type of element characterized by the presence of electrons in the *d* orbital
- \_\_\_\_\_ 3. type of element characterized by losing electrons easily
- \_\_\_\_\_ 4. ability of an atom to attract electrons when the atom is in a compound
- \_\_\_\_\_ 5. energy required to remove an electron from an atom

### Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- \_\_\_\_\_ 6. What is another name for the main group elements?
- |                              |                         |
|------------------------------|-------------------------|
| a. s block elements          | c. s & p block elements |
| b. p block elements elements | d. d block elements     |
- \_\_\_\_\_ 7. What is another name for the transition metals?
- |                              |                              |
|------------------------------|------------------------------|
| a. noble gases               | c. p block elements elements |
| b. s block elements elements | d. d block elements          |
- \_\_\_\_\_ 8. Who arranged the elements according to atomic mass and used the arrangement to predict the properties of missing elements?
- |                      |                     |
|----------------------|---------------------|
| a. Henry Moseley     | c. John Dalton      |
| b. Antoine Lavoisier | d. Dmitri Mendeleev |
- \_\_\_\_\_ 9. Which of the following categories includes the majority of the elements?
- |               |              |
|---------------|--------------|
| a. metalloids | c. metals    |
| b. liquids    | d. nonmetals |
- \_\_\_\_\_ 10. In which of the following sets is the symbol of the element, the number of protons, and the number of electrons given correctly?
- |                                 |                                    |
|---------------------------------|------------------------------------|
| a. In, 49 protons, 49 electrons | c. Cs, 55 protons, 132.9 electrons |
| b. Zn, 30 protons, 60 electrons | d. F, 19 protons, 19 electrons     |
- \_\_\_\_\_ 11. The atomic number of an element is the total number of which particles in the nucleus?
- |             |                          |
|-------------|--------------------------|
| a. neutrons | c. electrons             |
| b. protons  | d. protons and electrons |
- \_\_\_\_\_ 12. What element has the electron configuration  $1s^2 2s^2 2p^6 3s^2 3p^2$ ?
- |             |            |
|-------------|------------|
| a. nitrogen | c. silicon |
| b. selenium | d. silver  |
- \_\_\_\_\_ 13. Which subatomic particle plays the greatest part in determining the properties of an element?
- |             |                      |
|-------------|----------------------|
| a. proton   | c. neutron           |
| b. electron | d. none of the above |
- \_\_\_\_\_ 14. Which of the following elements is a transition metal?
- |           |              |
|-----------|--------------|
| a. cesium | c. tellurium |
| b. copper | d. tin       |

# Holt Chemistry The Periodic Table Chapter Test

**Marcel A. Müller**



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## **Table of Contents Holt Chemistry The Periodic Table Chapter Test**

1. Understanding the eBook Holt Chemistry The Periodic Table Chapter Test
  - The Rise of Digital Reading Holt Chemistry The Periodic Table Chapter Test
  - Advantages of eBooks Over Traditional Books
2. Identifying Holt Chemistry The Periodic Table Chapter Test
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Holt Chemistry The Periodic Table Chapter Test
  - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Chemistry The Periodic Table Chapter Test
  - Personalized Recommendations
  - Holt Chemistry The Periodic Table Chapter Test User Reviews and Ratings
  - Holt Chemistry The Periodic Table Chapter Test and Bestseller Lists
5. Accessing Holt Chemistry The Periodic Table Chapter Test Free and Paid eBooks
  - Holt Chemistry The Periodic Table Chapter Test Public Domain eBooks
  - Holt Chemistry The Periodic Table Chapter Test eBook Subscription Services
  - Holt Chemistry The Periodic Table Chapter Test Budget-Friendly Options

6. Navigating Holt Chemistry The Periodic Table Chapter Test eBook Formats
  - ePub, PDF, MOBI, and More
  - Holt Chemistry The Periodic Table Chapter Test Compatibility with Devices
  - Holt Chemistry The Periodic Table Chapter Test Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Holt Chemistry The Periodic Table Chapter Test
  - Highlighting and Note-Taking Holt Chemistry The Periodic Table Chapter Test
  - Interactive Elements Holt Chemistry The Periodic Table Chapter Test
8. Staying Engaged with Holt Chemistry The Periodic Table Chapter Test
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Holt Chemistry The Periodic Table Chapter Test
9. Balancing eBooks and Physical Books Holt Chemistry The Periodic Table Chapter Test
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Holt Chemistry The Periodic Table Chapter Test
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Holt Chemistry The Periodic Table Chapter Test
  - Setting Reading Goals Holt Chemistry The Periodic Table Chapter Test
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Holt Chemistry The Periodic Table Chapter Test
  - Fact-Checking eBook Content of Holt Chemistry The Periodic Table Chapter Test
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

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