

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

Holt Chemistry Section 21 Review Answer

Chao Zhang

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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Table of Contents Holt Chemistry Section 21 Review Answer

1. Understanding the eBook Holt Chemistry Section 21 Review Answer
 - The Rise of Digital Reading Holt Chemistry Section 21 Review Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Chemistry Section 21 Review Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Chemistry Section 21 Review Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Chemistry Section 21 Review Answer
 - Personalized Recommendations
 - Holt Chemistry Section 21 Review Answer User Reviews and Ratings
 - Holt Chemistry Section 21 Review Answer and Bestseller Lists
5. Accessing Holt Chemistry Section 21 Review Answer Free and Paid eBooks
 - Holt Chemistry Section 21 Review Answer Public Domain eBooks
 - Holt Chemistry Section 21 Review Answer eBook Subscription Services

- Holt Chemistry Section 21 Review Answer Budget-Friendly Options
- 6. Navigating Holt Chemistry Section 21 Review Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Holt Chemistry Section 21 Review Answer Compatibility with Devices
 - Holt Chemistry Section 21 Review Answer Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Chemistry Section 21 Review Answer
 - Highlighting and Note-Taking Holt Chemistry Section 21 Review Answer
 - Interactive Elements Holt Chemistry Section 21 Review Answer
- 8. Staying Engaged with Holt Chemistry Section 21 Review Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Chemistry Section 21 Review Answer
- 9. Balancing eBooks and Physical Books Holt Chemistry Section 21 Review Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Chemistry Section 21 Review Answer
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Holt Chemistry Section 21 Review Answer
 - Setting Reading Goals Holt Chemistry Section 21 Review Answer
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Holt Chemistry Section 21 Review Answer
 - Fact-Checking eBook Content of Holt Chemistry Section 21 Review Answer
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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