

## Chapter

## 6

The Chemistry of Life, *continued*

## Reinforcement and Study Guide

## Section 6.2 Water and Diffusion

*In your textbook, read about water and its importance.*

For each statement below, write **true** or **false**.

- \_\_\_\_\_ 1. In a water molecule, electrons are shared equally between the hydrogen atoms and oxygen atom.
- \_\_\_\_\_ 2. The attraction of opposite charges between hydrogen and oxygen forms a weak oxygen bond.
- \_\_\_\_\_ 3. Because of its polarity, water can move from the roots of a plant up to its leaves.
- \_\_\_\_\_ 4. Water changes temperature easily.
- \_\_\_\_\_ 5. Unlike most substances, water expands when it freezes.

Circle the letter of the choice that best completes the statement.

6. All objects in motion have  
a. potential energy.      b. heat energy.      c. kinetic energy.      d. random energy.
7. The first scientist to observe evidence of the random motion of molecules was  
a. Brown.      b. Darwin.      c. Mendel.      d. Hooke.
8. The net movement of particles from an area of higher concentration to an area of lower concentration is called  
a. dynamic equilibrium.      b. nonrandom movement.  
c. concentration gradient.      d. diffusion.
9. Diffusion occurs because of  
a. nonrandom movement of particles.      b. random movement of particles.  
c. a chemical reaction between particles.      d. chemical energy.
10. When a few drops of colored corn syrup are added to a beaker of pure corn syrup, the color will  
a. move from low concentration to high concentration.  
b. form a polar bond.  
c. start to diffuse.  
d. remain on the bottom of the beaker.
11. Diffusion can be accelerated by  
a. decreasing the pressure.      b. increasing the temperature.  
c. decreasing the movement of particles.      d. increasing the dynamic equilibrium.
12. When materials pass into and out of a cell at equal rates, there is no net change in concentration inside the cell. The cell is in a state of  
a. dynamic equilibrium.      b. metabolism.      c. imbalance.      d. inertia.
13. The difference in concentration of a substance across space is called  
a. dynamic equilibrium.      b. concentration gradient.  
c. diffusion.      d. Brownian movement.

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