

**LESSON**  
**7.5****Practice B***For use with pages 501–507***Match the expression with the logarithm that has the same value.**

1.  $\log \sqrt{2} + \log \sqrt{8}$

2.  $\log 4 - \log 10$

3.  $2 \log 4 - \log 2$

4.  $-3 \log \frac{1}{3}$

A.  $\log \frac{2}{5}$

B.  $\log 27$

C.  $\log 4$

D.  $\log 8$

**Use  $\log 4 = 0.602$  and  $\log 7 = 0.845$  to evaluate the logarithm.**

5.  $\log 28$

6.  $\log \frac{7}{4}$

7.  $\log 16$

8.  $\log 49$

9.  $\log \frac{1}{4}$

10.  $\log \frac{49}{64}$

**Expand the expression.**

11.  $\log_3 3x$

12.  $\log \frac{2x}{5}$

13.  $\log_7 x^2y$

14.  $\log_2 \frac{x^2}{4}$

15.  $\ln \sqrt[3]{xy}$

16.  $\log 5 \sqrt[3]{x}$

17.  $\ln \frac{1}{2x^2}$

18.  $\log_6 \frac{2x^2}{3}$

19.  $\log_6 \frac{xy^2}{\sqrt{x}}$

**Condense the expression.**

20.  $\log_3 4 + \log_3 2 + \log_3 2$

21.  $\log 4 + 3 \log x + \log y$

22.  $\log 3 + \frac{1}{2} \log x - \log 5$

23.  $2 \ln x - \ln 3 + \ln 6$

24.  $3 \log x + \log 4 - \log x - \log 6$

25.  $3 \ln(x + 1) - 2 \ln y + \ln y + \ln 2$

**Use the change-of-base formula to evaluate the logarithm.****Round your result to three decimal places.**

26.  $\log_7 12$

27.  $\log_4 112$

28.  $\log_5 1.25$

29.  $\log_{2.2} 22$

30.  $\log_{4.2} 18.1$

31.  $\log_{3.3} 0.0005$

**In Exercises 32–34, use the following information.**

**Henderson-Hasselbalch Formula** The pH of a patient's blood can be calculated using the Henderson-Hasselbalch Formula,  $\text{pH} = 6.1 + \log \frac{B}{C}$ , where  $B$  is the concentration of bicarbonate and  $C$  is the concentration of carbonic acid. The normal pH of blood is approximately 7.4.

**32.** Expand the right side of the formula.**33.** Find the pH of blood that has bicarbonate concentration of 38 and carbonic acid concentration of 2.0?**34.** Is the pH in Exercise 33 above normal or below normal?**Algebra 2**

Chapter 7 Resource Book

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**Holt McDougal, Ron Larson**



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