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MPUMALANGA PROVINCE
REPUBLIC OF SOUTH AFRICA

**PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 1 FORMAL EXPERIMENT - PART 2
VERIFICATION OF BOYLE'S LAW
MARKING GUIDELINES**

- 1.1 • Aim stated correctly. ✓
e.g. To determine the relationship between the volume of a fixed amount of gas and the pressure on the gas when temperature remains constant.

(1)

- 1.2 • All apparatus listed ✓

1. Boyle's Apparatus
2. Pump

- All steps of the method correct and logically stated. ✓✓ correct tense used. ✓✓

e.g.

1. The tap on the Boyle's Law apparatus was opened and then the pump was connected.
2. The apparatus was pumped to the pressure.
3. The oil was allowed to settle for a few minutes.
4. A table was used to record the volume reading on the tube and the corresponding pressure reading on the Bourdon pressure gauge.
5. The air tap was slightly opened and quickly closed to let a little air escape from the system. The oil was allowed to settle for a few minutes and then the pressure and volume readings were taken.
6. 5 different readings were recorded in the table.

(5)

2. Observations and Recording of results:

Learners will use results obtained during experiment to complete the table below.

(6)

Volume (V) cm ³	Pressure (p) kPa		pV (✓)

Marking guidelines:

- ✓ for correct values of the volume
- ✓ for correct values of the pressure
- ✓✓ for correct values of

Grade Prescribed Experiment 2 Question Paper And Memorandum

Tobias Bleicker



Grade Prescribed Experiment 2 Question Paper And Memorandum:

Grade Prescribed Experiment 2 Question Paper And Memorandum Book Review: Unveiling the Magic of Language

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Table of Contents Grade Prescribed Experiment 2 Question Paper And Memorandum

1. Understanding the eBook Grade Prescribed Experiment 2 Question Paper And Memorandum
 - The Rise of Digital Reading Grade Prescribed Experiment 2 Question Paper And Memorandum
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade Prescribed Experiment 2 Question Paper And Memorandum
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade Prescribed Experiment 2 Question Paper And Memorandum
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade Prescribed Experiment 2 Question Paper And Memorandum
 - Personalized Recommendations
 - Grade Prescribed Experiment 2 Question Paper And Memorandum User Reviews and Ratings
 - Grade Prescribed Experiment 2 Question Paper And Memorandum and Bestseller Lists

5. Accessing Grade Prescribed Experiment 2 Question Paper And Memorandum Free and Paid eBooks
 - Grade Prescribed Experiment 2 Question Paper And Memorandum Public Domain eBooks
 - Grade Prescribed Experiment 2 Question Paper And Memorandum eBook Subscription Services
 - Grade Prescribed Experiment 2 Question Paper And Memorandum Budget-Friendly Options
6. Navigating Grade Prescribed Experiment 2 Question Paper And Memorandum eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade Prescribed Experiment 2 Question Paper And Memorandum Compatibility with Devices
 - Grade Prescribed Experiment 2 Question Paper And Memorandum Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade Prescribed Experiment 2 Question Paper And Memorandum
 - Highlighting and Note-Taking Grade Prescribed Experiment 2 Question Paper And Memorandum
 - Interactive Elements Grade Prescribed Experiment 2 Question Paper And Memorandum
8. Staying Engaged with Grade Prescribed Experiment 2 Question Paper And Memorandum
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade Prescribed Experiment 2 Question Paper And Memorandum
9. Balancing eBooks and Physical Books Grade Prescribed Experiment 2 Question Paper And Memorandum
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade Prescribed Experiment 2 Question Paper And Memorandum
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Grade Prescribed Experiment 2 Question Paper And Memorandum
 - Setting Reading Goals Grade Prescribed Experiment 2 Question Paper And Memorandum
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Grade Prescribed Experiment 2 Question Paper And Memorandum
 - Fact-Checking eBook Content of Grade Prescribed Experiment 2 Question Paper And Memorandum
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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