



MERCY JUNIOR COLLEGE
Grade 11
Physical Science
Finals Exam

NAME: _____

YEAR LEVEL & SECTION: _____

TEACHER'S NAME: _____

PROCTOR'S NAME: _____

1. MULTIPLE CHOICE

INSTRUCTION: Encircle the letter of the correct answer. (1 POINT EACH)

- It is a subatomic particle, symbol e^- or β^- , with a negative elementary electric charge.
 - Neutron
 - Electron
 - Proton
 - Atom
- It is a subatomic particle, symbol p or p^+ , with a positive electric charge of $+1e$ elementary charge and mass slightly less than that of a neutron.
 - Neutron
 - Electron
 - Proton
 - Atom
- It is a subatomic particle, symbol n or n^0 , with no net electric charge and a mass slightly larger than that of a proton.
 - Neutron
 - Electron
 - Proton
 - Atom
- Is this an example of ionic bonding?



carbon dioxide

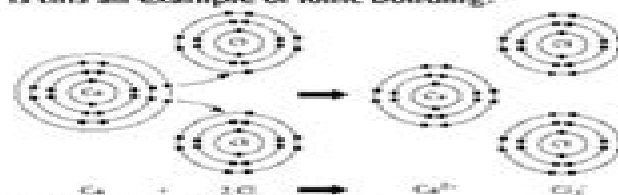
- Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- This is just an orbital model.

- Is this an example of ionic bonding?



lithium fluoride

- Yes, it is ionic bonding.
 - No, it is of another type of chemical bonding.
 - There is no chemical bonding present.
 - This is just an orbital model.
- In atomic physics and quantum chemistry, the electron configuration is the distribution of electrons of an atom or molecule in atomic or molecular orbitals.
 - Electron Configuration
 - Periodic Table of Elements
 - Orbital Models
 - Energy Levels
 - Is this an example of ionic bonding?



- Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- This is just an orbital model.

- It is a chemical rule of thumb that reflects observation that atoms of main-group elements tend to combine in such a way that each atom has eight electrons in its valence shell, giving it the same electronic configuration as a noble gas.
 - Octet Rule
 - Valence Electrons
 - Atomic Orbitals
 - Chemical Bonds
- It is the smallest constituent unit of ordinary matter that has the properties of a chemical element.
 - Neutron
 - Electron
 - Proton
 - Atom
- It is a type of chemical bond that involves the electrostatic attraction between oppositely

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Marco Cascella



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