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# Fellowes C 22user Guide

**Lynda J. Carpenito**



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$$\text{MnO}_4^- + 4\text{H}^+ + 3\text{e}^- \rightarrow \text{MnO}_2 + 2\text{H}_2\text{O}$$

$$\text{MnO}_4^- + 2\text{H}^+ + \text{e}^- \rightarrow \text{MnO}_4^{2-} + \text{H}_2\text{O}$$

$$\text{MnO}_4^- + \text{e}^- \rightarrow \text{MnO}_4^{2-}$$

basic

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